Possibilities and challenges for PFAS analysis

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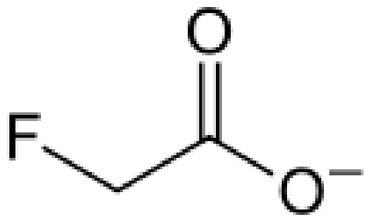
MTM Research Centre, Örebro University, Sweden

PFAS Contaminated sites risk-assessment 2020-11-09

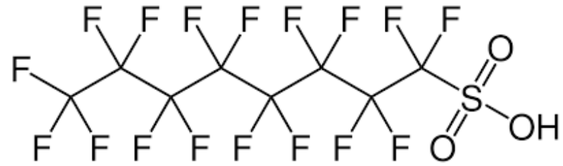
Different forms of fluorine



most common form found in nature



Natural occuring organofluorine



Antropogenic organofluorine

E.g.

- CaF_2
- *NaF*
- Na_2PO_3F

E.g.

- *Trifluoromethane*
- *Fluoroacetate*
- *Fluorinated fatty acids*

E.g.

- *PFAS*
- *Pharmaceuticals*
- *Agrochemicals*

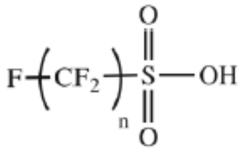
Per- and polyfluoroalkyl substances (PFAS)

4730 PFASs (CAS-numbers, OECD definition 2018)

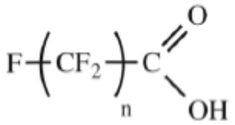


Perfluoroalkyl acids (PFAA)

Examples:



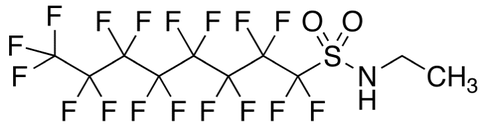
PFSA



PFCA

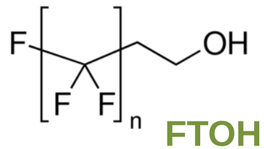
Precursors

Examples of precursors to PFSA:

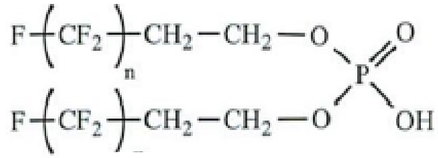


Et-FOSA

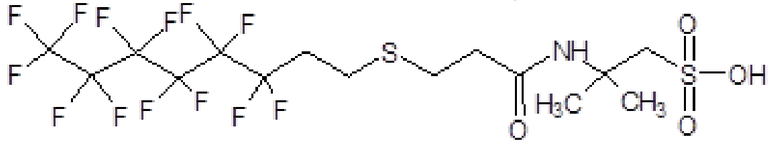
Example of precursors to PFCA:



FTOH



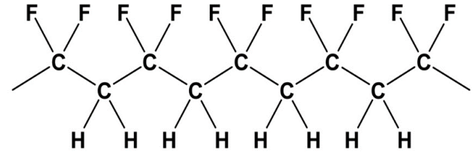
diPAP

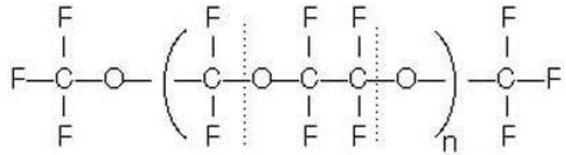


6:2 FTSAS

Others

Polymers:



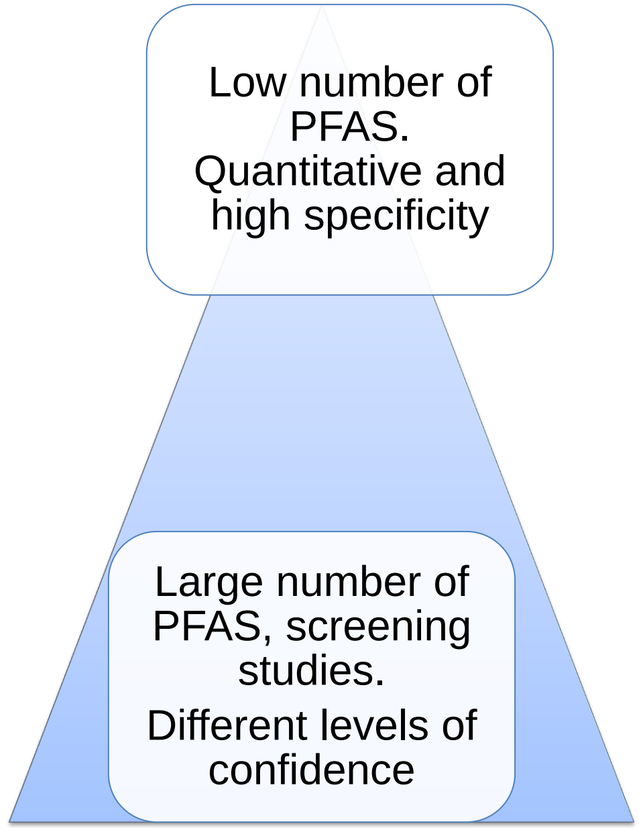


What are being monitored, and why?

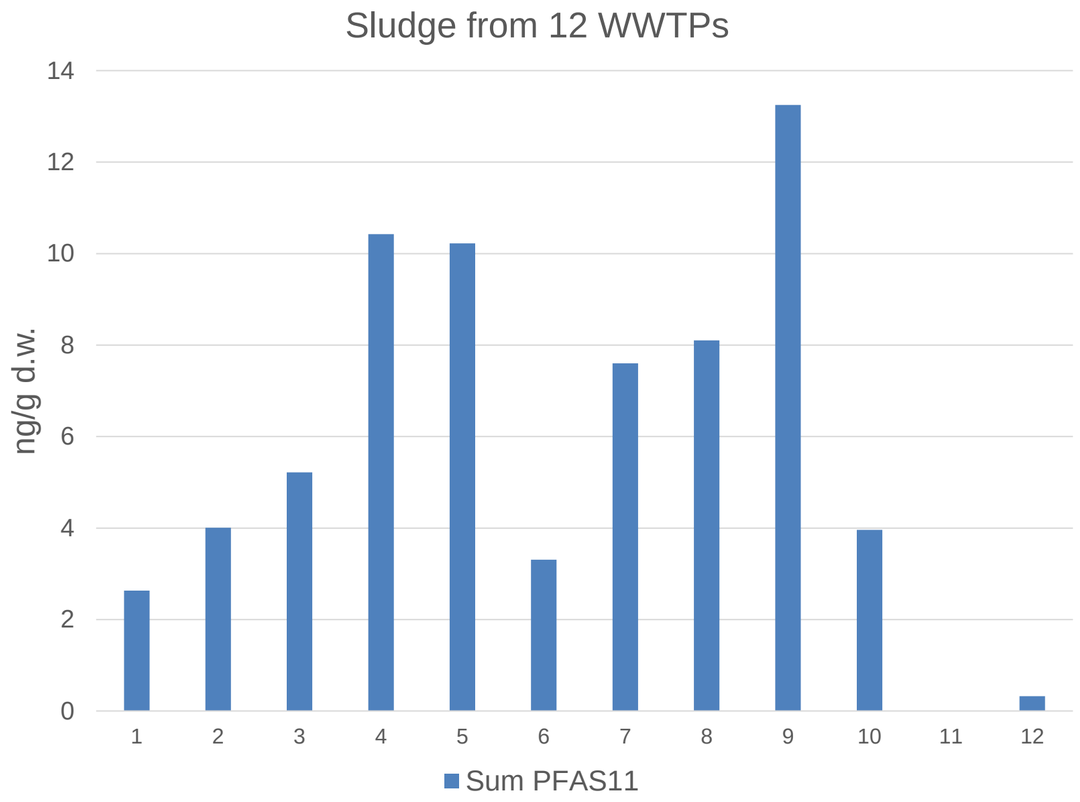
Why?

- Comply to regulations
- Study health effects and impacts
- Study removal efficiency in WWTP, drinking water, remediation,…
- Study environmental distribution and fate
- Occurrence of (new) PFAS in products, environment, humans,…

What?



Example from waste water treatment



∑PFAS11:

PFBA

PFPeA

PFHxA

PFHpA

PFOA

PFNA

PFDA

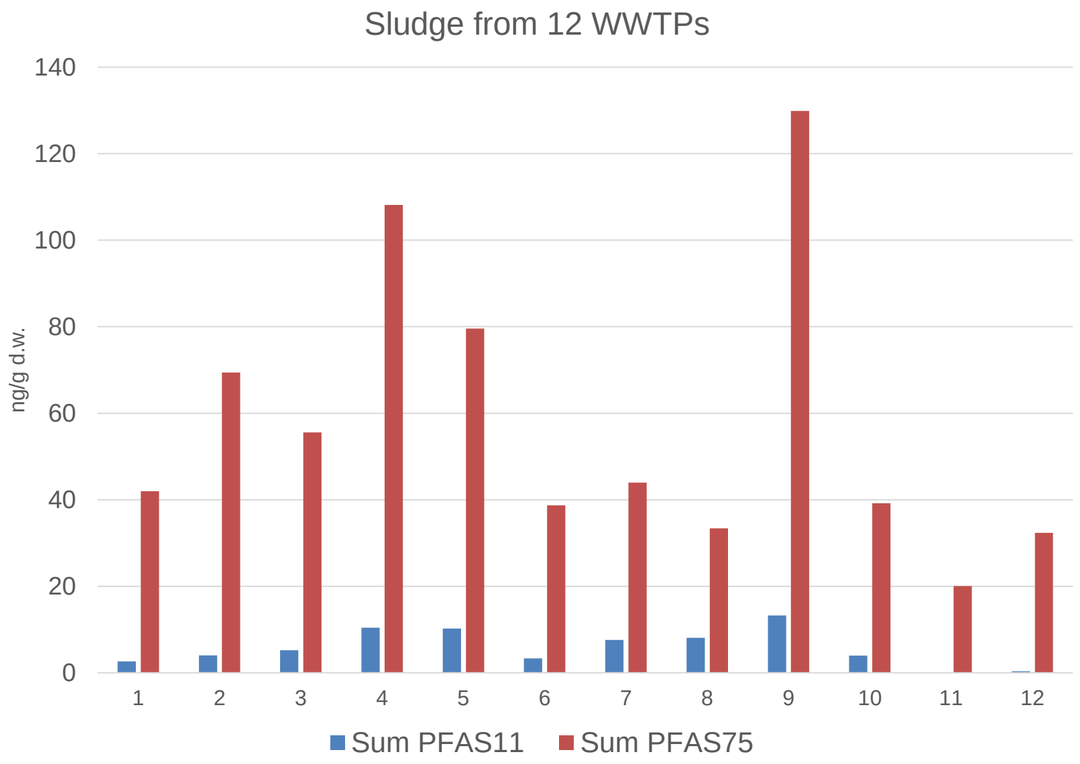
PFBS

PFHxS

PFOS

6:2 FTSA

Example from waste water treatment



Are there more PFAS?

∑PFAS11:

PFBA

PFPeA

PFHxA

PFHpA

PFOA

PFNA

PFDA

PFBS

PFHxS

PFOS

6:2 FTSA

∑PFAS75:

PFCAs

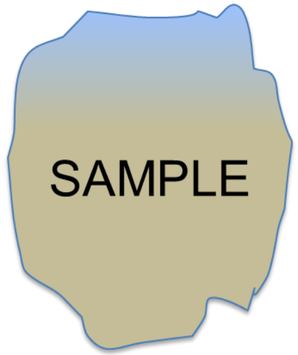
PFSAs

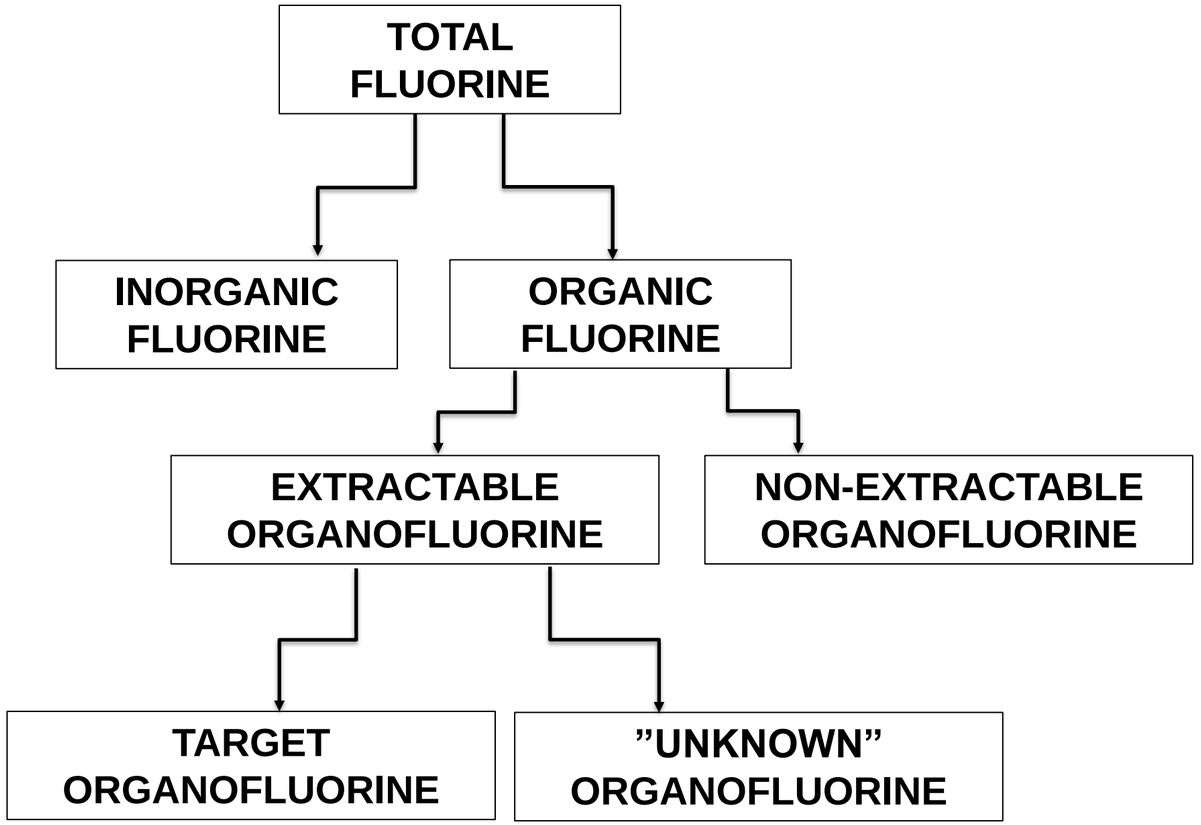
Precursors: PAPs, FTSAs, FTUCA, FTCA FOSA, FOSAA, diSamPAP,

PFPA/PFPiA

ADONA, HFPO-DA, PFECHS, F53B

Analytical possibilities

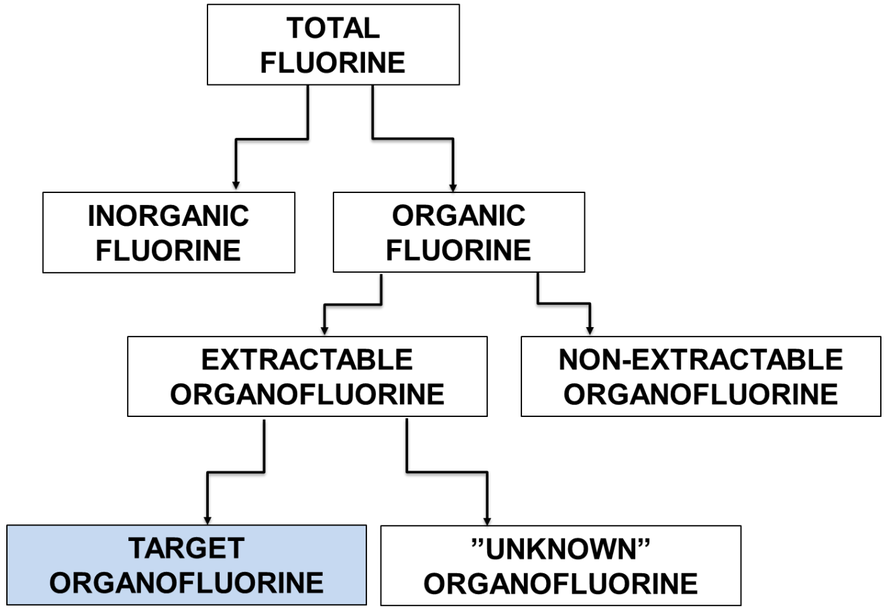




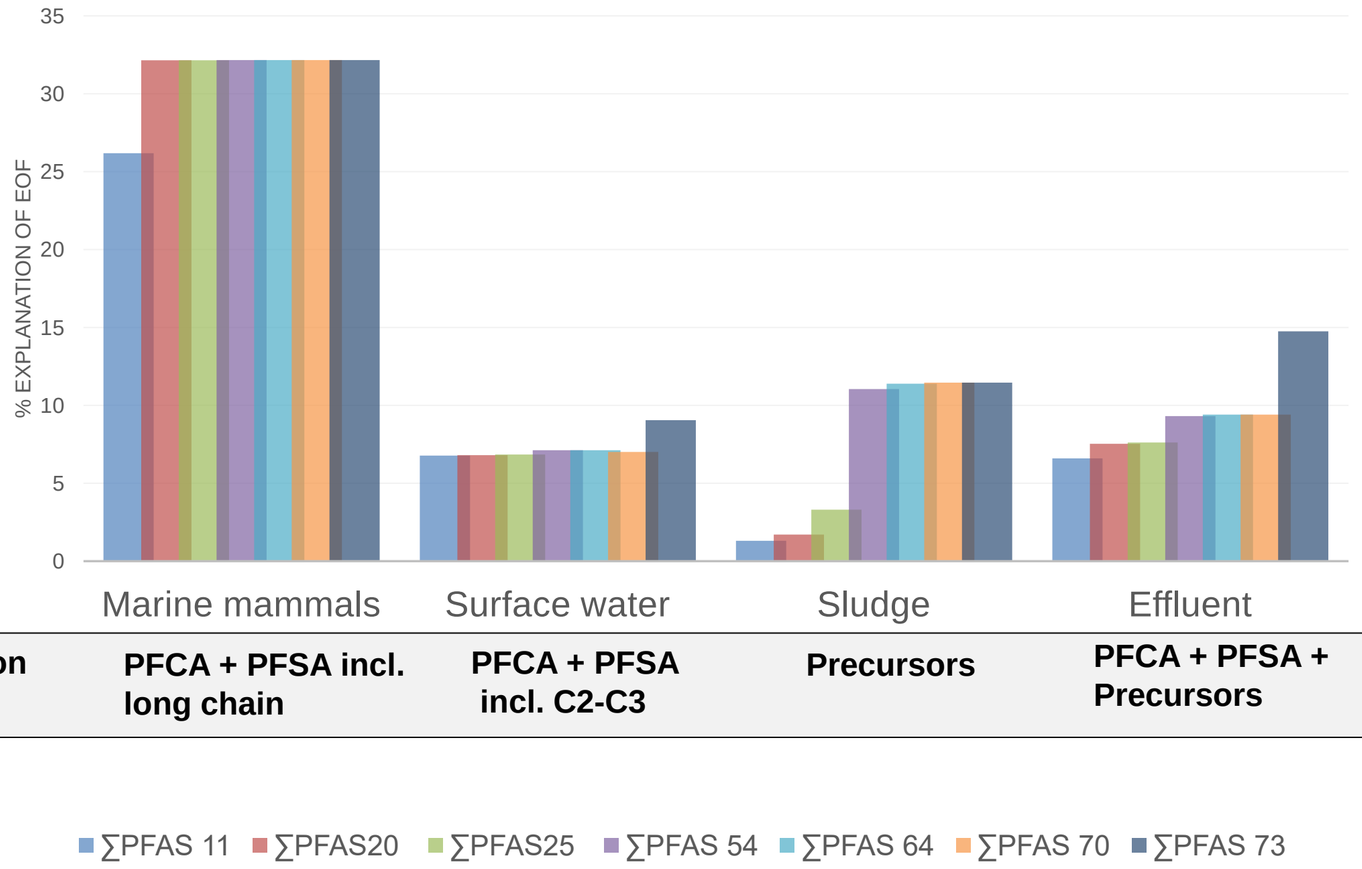
Target analysis

PFAA and selected precursors:

- Sensitive (ppt-levels) and specific
- Routine measurement of ~40 PFAS
- Further challenges:
    - increase physicochemical range to be more inclusive
    - commercial standards
- Is it possible to target all PFAS?



Profile analysis



Largest contribution to Organofluorine:	**PFCA + PFSA incl. long chain**	**PFCA + PFSA incl. C2-C3**	**Precursors**	**PFCA + PFSA + Precursors**

Profiles in environmental compartments

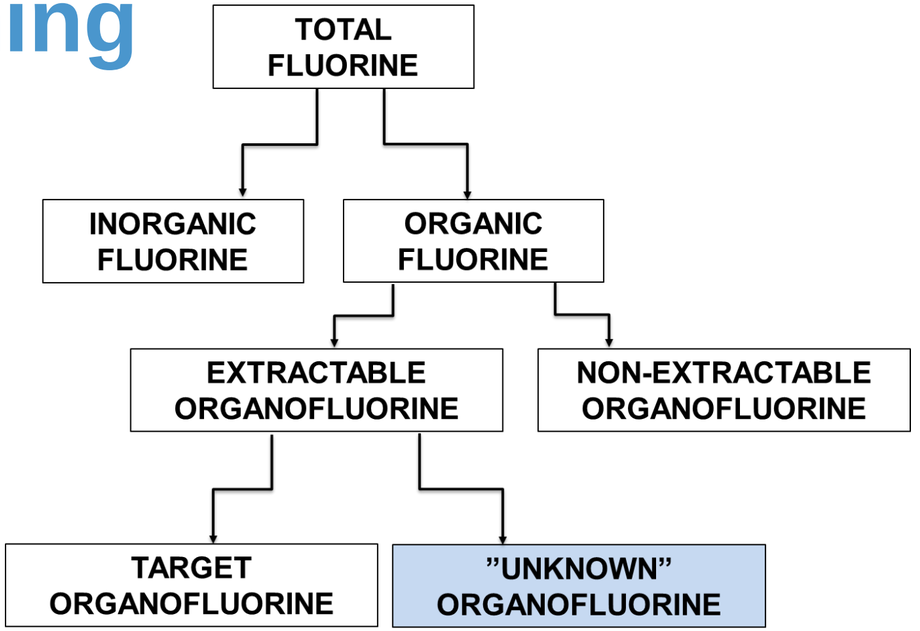
- surface water
- soil
- ground water
- Biota

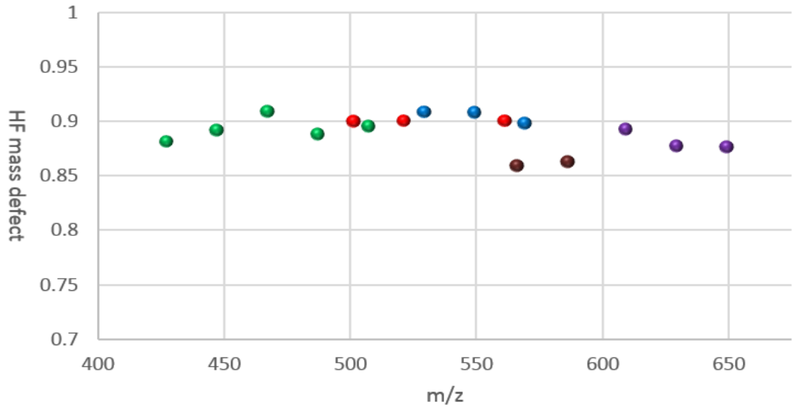
Profiles from different contamination sources :

- historical AFFF
- recent AFFF
- ski wax
- paper treatment
- textile treatments
- etc.

Suspect and nontarget screening

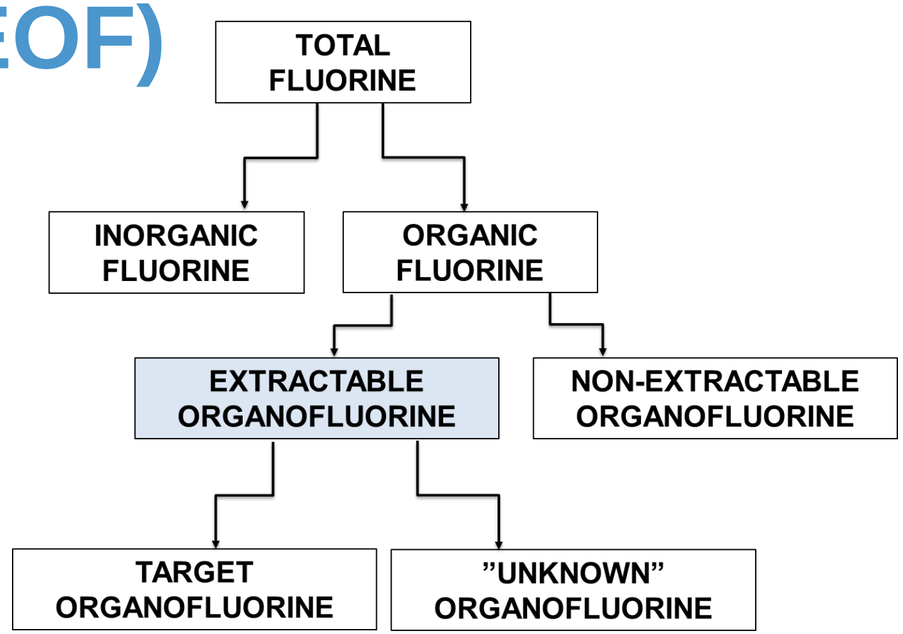
- Increasingly used to detect known unknowns (suspects), or unknowns (nontargeted)
- Limited possibilities for quantitative analysis
- Automation challenges for data mining and reduction
  - mass defects
  - fragment flagging
  - suspect lists



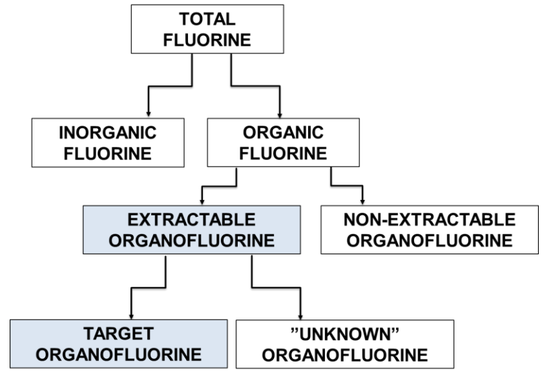


Extractable organofluorine (EOF)

- Measure the elemental fluorine in an extract
- Possibilities to separate and fractionate different classes
    - High selectivity towards organofluorine after removing inorganic fluorine
- Results depends on the extraction method



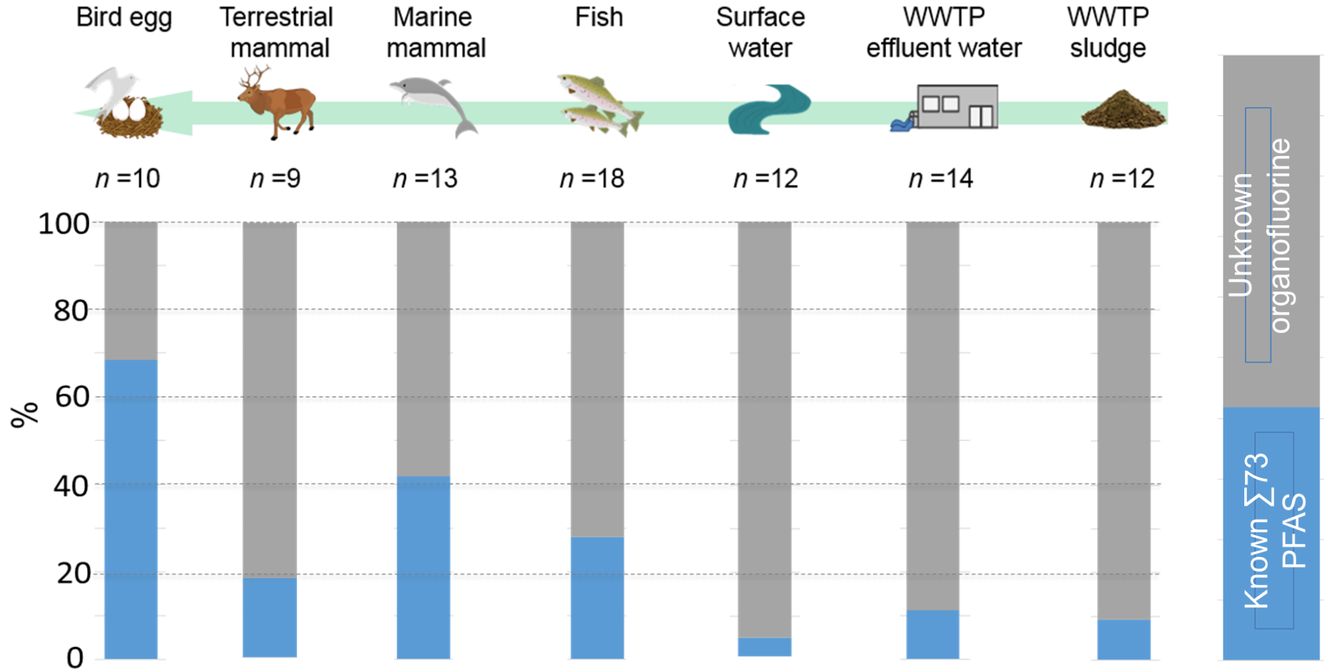
Fluorine mass balance to show possible unidentified PFAS



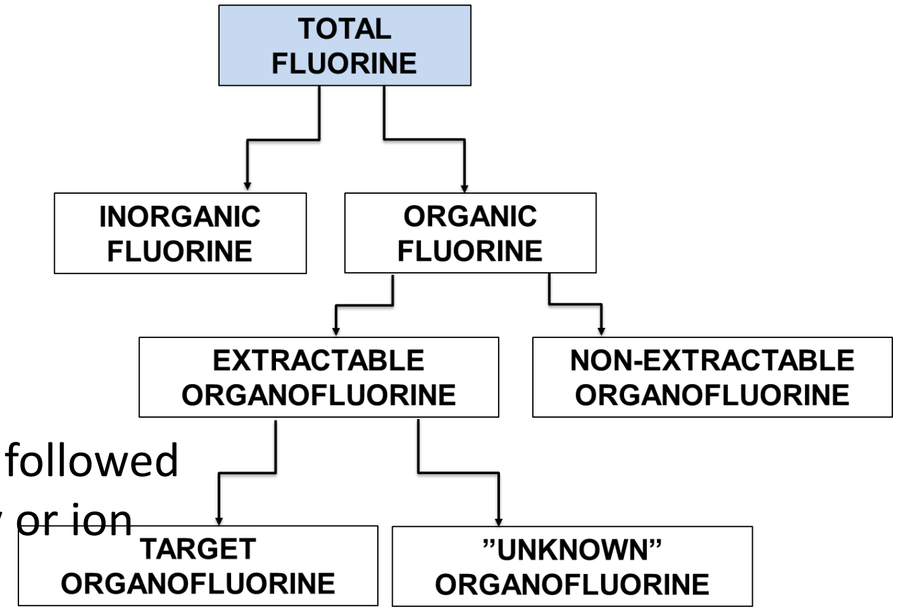
Comparing EOF with target PFAS can indicate presence of unknowns

Mass balance = C_F target / C_F EOF

- no information on identity of unknowns
- measuring F is less sensitive than target PFAS



Total fluorine



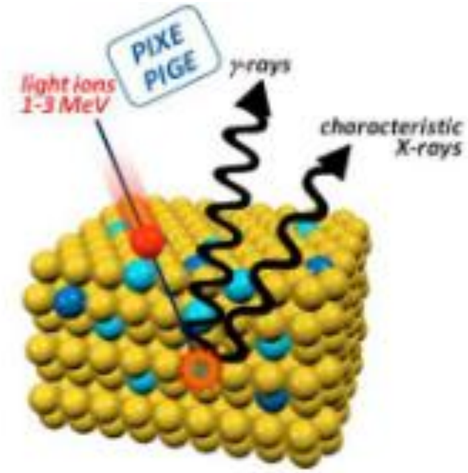
Indicates the presence of fluorine

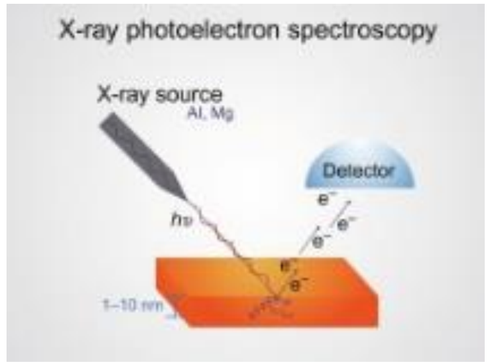
- Combustion
  - Pyrolysis (1000-1100 ° C) and thermal oxidation followed by fluoride measurement by ion chromatography or ion selective electrode
- Surface analysis
  - Particle Induced Gamma-ray Emission (PIGE) spectroscopy
    - surface 100−250 µm
  - X-ray photoelectron spectroscopy (XPS)
    - surface ≤0.1 µm

Total fluorine

- Combustion
  - Inclusive for all elemental fluoride
  - Less sensitive (0.1-0.5 ppm F) than targeted analyses
- Surface analysis
  - Particle Induced Gamma-ray Emission (PIGE) spectroscopy
    - High-throughput
    - Non-destructive
    - Less sensitive (50 nmol F/cm^2) than targeted analyses
  - X-ray photoelectron spectroscopy (XPS)
    - High-throughput
    - Widely-available instrumentation
    - Identification of perfluoroalkyl moiety
    - Detection limit 1%

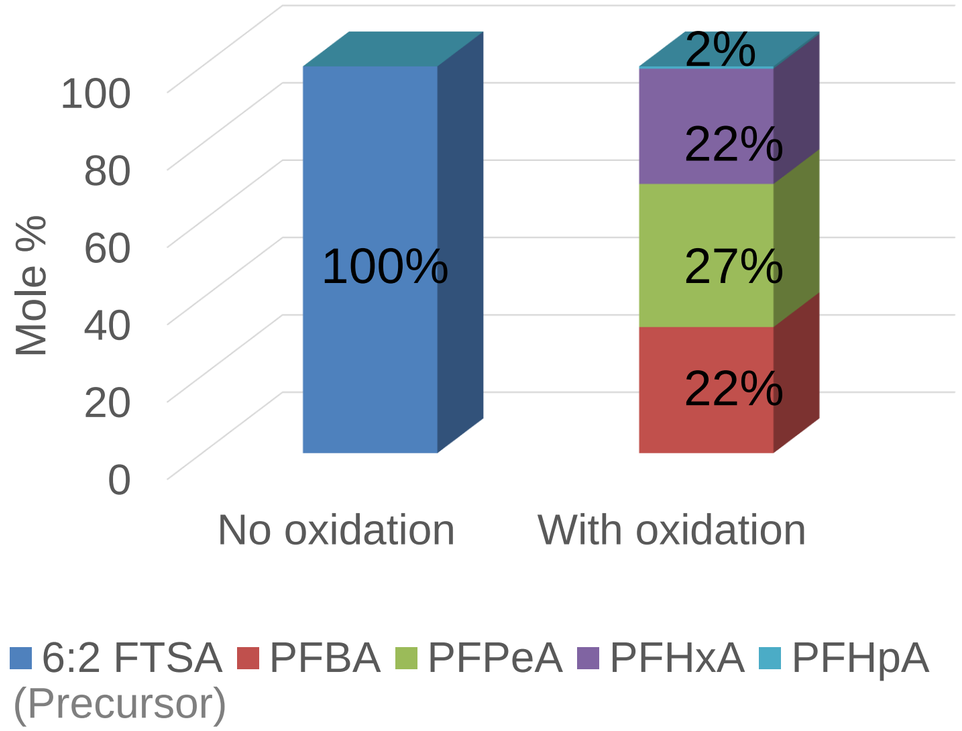






Total oxidizable precursor (TOP) assay

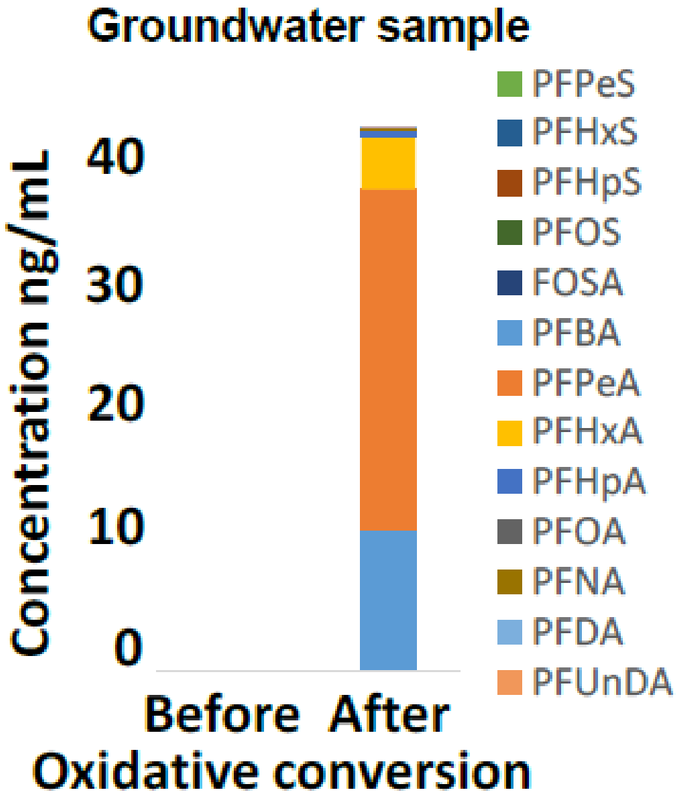
Chemical reaction of a *sample** or a *sample extract* converts precursors to perfluoroalkyl acids



*mainly aqueous samples

Total oxidizable precursor (TOP) assay

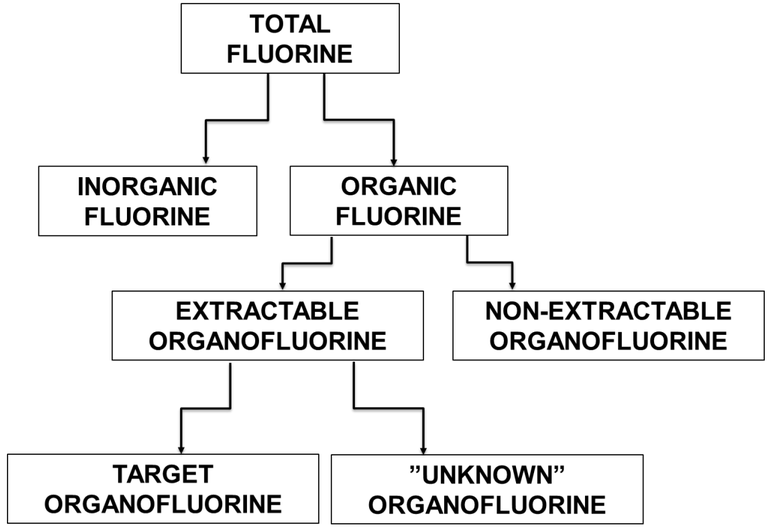
- Sensitive analysis (ppt-levels) after successful reaction
- Reaction products depends on structure of precursors (e.g. functional group and chain lengths)
- Reaction results depends on precursor stability towards hydroxyl oxidation



Conclusion

Available inclusive methods for total fluorine, extractable organofluorine, and oxidizable precursors can act as triggers for further actions

- develop indicator values for background - and different contamination levels
- describe the method`s performances for different matrices and circumstances



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